Preparations of Trisubstituted Hydrazines and Pyrazolidines from N-(1-Benzotriazolylalkyl)hydrazines

Alan R. Katritzky,* Guofang Qiu, and Baozhen Yang

Center for Heterocyclic Compounds, Department of Chemistry, University of Florida, Gainesville, Florida 32611–7200

Received May 22, 19978

Reactions of N,N-disubstituted hydrazines with benzotriazole and aldehydes give N-(α -benzotriazoylalkyl)-N,N-disubstituted hydrazines which on treatment with organometallic reagents form N-alkyl-N,N-disubstituted hydrazines in good yields. N-(Benzotriazolylalkyl)-N,N-disubstituted hydrazines and electron-rich olefins, in the presence of zinc bromide catalyst, generate N,N-disubstituted pyrazolidines in moderate to good yields.

Introduction

Hydrazines are important because of their many applications in organic synthesis and industry. Trisubstituted hydrazines are usually prepared by (i) carbanion addition to the N=N bond of an azo compound using a silyl ester enolate¹ or a lithium amide enolate².³ or of an alkyl- or aryllithium,⁴ routes which are limited by the type and accessibility of the appropriate nucleophiles; (ii) the alkylation of mono- or disubstituted hydrazines, which rarely gives good yields;⁵ (iii) electrophilic aminations of amines;^{6,7} and (iv) additions of organometallic reagents to hydrazones.⁸

Pyrazolidines have been widely investigated as fungicides, herbicides, antiinflammatories, antibacterials, anesthetics, and anticonvulsants. General methods for the synthesis of pyrazolidines include (i) 1,3-dipole azomethine—imine additions to carbon—carbon double bonds; (ii) reductions of pyrazolines, pyrazoles, pyrazolones, and pyrazolium salts; Als and (iii) cyclizations of hydrazines with halogenated compounds.

Benzotriazole has been extensively used as a synthetic auxiliary in organic synthesis. $^{16-18}\,$ N-Alkylbenzotriazoles with an electron-donor group at the $\alpha\text{-position}$ of the N-alkyl group can undergo facile replacement of the

- Abstract published in Advance ACS Abstracts, October 1, 1997.
 (1) Gennari, C.; Colombo, L.; Bertolini, G. J. Am. Chem. Soc. 1986,
- (2) Evans, D. A.; Britton, T. C.; Dorow, R. L.; Dellaria, J. F. *J. Am. Chem. Soc.* **1986**, *108*, 6395.
 - (3) Trimble, L. A.; Vederas, J. C. J. Am. Chem. Soc. 1986, 108, 6397.
- (4) Katritzky, A. R.; Wu, J.; Verin, S. V. Synthesis 1995, 651. (5) Askani, R.; Taber, D. F. In Comprehensive Organic Synthesis; Trost, B. M.; Fleming, I., Eds.; Pergamon Press: Oxford, 1991; Vol. 6, 116
- (6) Boche, G.; Sommerlade, R. H.; Bosold, F. *Angew. Chem., Int. Ed. Engl.* **1986**, *25*, 562.
- (7) Linke, K.-H.; Turley, R. Z. Naturforsch, B **1973**, 28, 829.
- (8) Huddleston, P. R.; Coutts, I. G. C. In Comprehensive Organic Functional Group Transformations; Katritzky, A. R., Meth-Cohn, O., Rees, C. W., Eds.; Pergamon: Oxford, 1995; Vol. 2, p 371. Enders, D.; Lochtman, P.; Rabb, C. Synlott 1996, 136
- Lochtman, R.; Raabe, G. Synlett **1996**, 126. (9) Searching MEDLINE EXPRESS from 1966 to 1996, 59 records were found.
- (10) Elguero, J. In *Comprehensive Heterocyclic Chemistry*, Katritzky, A. R.; Rees, C. W., Eds.; Pergamon Press: New York, 1984; Vol. 5, p 297.
- (11) Huisgen, R.; Grashey, R.; Laur, P.; Leitermann, H. Angew. Chem. 1960, 72, 416.
 - (12) Oppolzer, W. *Tetrahedron Lett.* **1970**, 2199.
 - (13) Sun, B.; Adachi, K.; Noguchi, M. *Tetrahedron* **1996**, *52*, 901.
- (14) Behr, L. C.; Fusco, R.; Jarboe, C. H. In *Chemistry of Heterocyclic Compounds: Pyrazoles, Pyrazolines, Pyrazolidines, Indazoles and Condensed Rings*; Wiley, R. H., Ed.; Interscience Publishers: New York, 1967; p 279.
- (15) Banuelos, L. A.; Cuadrado, P.; Gonzalez-Nogal, A. M.; Lopez-Solera, I.; Pulido, F. J.; Raithby, P. R. *Tetrahedron* **1996**, *52*, 9193.

benzotriazolyl (Bt) group by nucleophiles such as Grignard reagents or electron-rich alkenes. We now report new preparations of (i) trisubstituted hydrazines in high yields by the reactions of N-(1-benzotriazolylalkyl)-N,N-disubstituted hydrazines with Grignard reagents or organozinc reagents and (ii) substituted pyrazolidines by the reactions of N-(1-benzotriazolylalkyl)-N,N-diarylhydrazines with electron-rich alkenes. We recently reported the preparation of trisubstituted and tetrasubstituted hydrazines in yields of 32–57% by additions of α -lithio-1-alkylbenzotriazoles to the N=N bond of azobenzenes, followed by substitution of the benzotriazole residue by Grignard reagents. The present investigations comprise a significant generalization and extension of this methodology.

Results and Discussion

Synthesis of N-(1-Benzotriazolylalkyl)-N,N-di**substituted Hydrazines.** *N*-(1-Benzotriazolylalkyl)-*N,N*-diphenylhydrazines **2a**-**d** were prepared in good yields from N,N-diphenylhydrazine (1a), benzotriazole, and the appropriate aldehyde in methylene chloride at rt in the presence of molecular sieves as dehydrating agent. These reactions gave mixtures of benzotriazol-1yl and benzotriazol-2-yl isomers, and these two isomers are in equilibrium in solution.20 The ratio of Bt-1 and Bt-2 isomers was 8 to 1, according to the integration of the ¹H NMR spectrum recorded immediately after dissolving the solid in CDCl₃. When the ¹H NMR spectrum of the same sample was recorded 30 min later, the ratio had changed to 2 to 1. Considering that both Bt-1 and Bt-2 groups are good leaving groups, the mixture of the two isomers was directly used for further reactions without separation.

1,4-Diphenyl-4-[(benzotriazol-1-yl)methyl]semicarbazide (**2e**) and *N*-phenyl-*N*-[(benzotriazol-1-yl)methyl]-*N*-propionylhydrazine (**2f**) were prepared from corresponding hydrazines, benzotriazole, and paraformaldehyde under reflux in benzene, with a Dean—Stark trap to remove the water produced, and both of them gave only

⁽¹⁶⁾ Katritzky, A. R.; Rachwal, S.; Hitchings, G. J. *Tetrahedron* **1991**, *47*, 2683.

⁽¹⁷⁾ Katritzky, A. R.; Lan, X.; Fan, W.-Q. Synthesis **1994**, 445. (18) Katritzky, A. R.; Lan, X.; Yang, J. Z.; Denisko, O. V. Chem.

⁽¹⁸⁾ Katritzky, A. R.; Lan, X.; Yang, J. Z.; Denisko, O. V. *Chen Rev.* **1997**, submitted.

⁽¹⁹⁾ Katritzky, A. R.; Verin, S. V.; Yang, B. *Org. Prepr. Proced. Int.* **1996**, *28*, 97.

⁽²⁰⁾ Katritzky, A. R.; Fan, W.-Q. *J. Org. Chem.* **1990**, *55*, 3205.

Scheme 1

$$\begin{array}{c|c} & & & \\ \hline & & \\ \hline X & & \\ \hline &$$

For description of R, R¹, and R², see Table 1.

Method A: CH2Cl2/rt, 3 Å molecular sieves for 2a-d. Method B: benzene/reflux, Dean-Stark trap for 2e-f.

Table 1. Preparation of N-(1-Benzotriazolylalkyl)-N,N'-disubstituted **Hydrazines 2**

compd	R ¹	R	time (h)	yield (%)	mp (°C)	Bt-1/ Bt-2 ^a
2a	Ph	Н	24	88	137-138	97:3
2b	Ph	Et	120	69	127 - 132	5:2
2b	Ph	n-C ₃ H ₇	120	64	122 - 126	2:1
2d	Ph	Cyclohexyl	120	70	153 - 156	2:1
2e	PhNHCO	Η̈́	48	61^{b}	183 - 184	
2f	C_2H_5CO	Н	48	61^{b}	157 - 158	

^a The ratio of Bt-1/Bt-2 observed just after concentration. ^b In refluxing benzene with Dean-Stark trap.

Table 2. Preparation of Trisustituted Hydrazines 3

compd	R	\mathbb{R}^2	yield (%)	mp (°C)
3a	Н	Et	79	oil
3 b	Et	Et	81	69 - 70
3c	n-C ₃ H ₇	Et	80	59 - 60
3d	cyclohexyl	Et	70	117 - 119
3e	n-C ₃ H ₇	Ph	68	117 - 118
3f	H	Allyl	99	oil
3g	n-C ₃ H ₇	Allyl	91	oil
3h	H	2-furyl	62	59 - 60
3i	H	2-thiophenyl	53	60 - 62

Bt-1 isomers. The results of these reactions are listed in Scheme 1 and Table 1.

Preparation of Trisubstituted Hydrazines 3. N-(1-Benzotriazolylalkyl)-N,N-diphenylhydrazines 2a-d reacted with Grignard reagents at 0 °C followed by warming to room temperature to give trisubstituted hydrazines 3a-g in good yields. Treatments of compound 2a with 2-furyl and 2-thiophenyl organozinc reagents gave 3h and 3i in 62% and 53% yields, respectively (Table 2).

Preparation of Substituted Pyrazolidines 4 and **5.** We have reported reactions of N-(1-benzotriazolylalkyl)aniline both with electron-rich²¹ and unactivated alkenes in the presence of Lewis acid^{22,23} to give tetrahydroquinolines. In contrast, we now find that treatment

Scheme 2 2a,e,f CH₂Cl₂ CH₂Cl₂ ZnBr₂ ZnBr₂ 5a-d ÷а $(4)\bar{H}$ 5b-d 4a-c R1 R1 Ph 4a 2-Pyrrolidonyl **PhNHCO** Ph 5c 2 4h N-Methylacetamidyl C₂H₅CO 5d PhNHCO OEt

Table 3. Synthesis of Pyrazolidines 4 and 5

compd	\mathbb{R}^1	X (for 4) or <i>n</i> (for 5)	time (h)	yield (%)	mp (°C)
4a	Ph	2-pyrrolidonyl	30	53	107-109
4b	Ph	<i>N</i> -methylacetamidyl	24	43	142-143
4c	PhNHCO	OEt	50	60	118-120
5a	Ph	n = 1	24	84	139 - 142
5b	Ph	n = 2	24	83	168 - 171
5c	PhNHCO	n = 2	48	60	221-224
5 d	C_2H_5CO	n = 2	48	39	124 - 126

of N-(1-benzotriazolylmethyl)-N,N-disubstituted hydrazines 2a with electron rich alkenes in the presence of Lewis acids gives pyrazolidines **4a**,**b**. Rather than the ortho-carbon of the phenyl ring, the nitrogen atom in hydrazines 2 readily attacks the cationic intermediates 6 formed from starting material 2 and the alkene (Scheme 2 and Table 3). In these reactions, $2 \rightarrow 4$, 2 equiv of alkene was needed because 1 equiv was consumed by the benzotriazolyl anion which is formed simultaneously.

Treatment of 2a with cyclic vinyl ethers gave cis-fused bicyclic compounds 5a and 5b exclusively; the cis relationship of H-3 and H-4 was determined by NOE experiments and the coupling H-3/H-4 constants in the ¹H NMR. These are $\hat{6}.1$ Hz for 5a and 1.7 Hz for 5b, consistant with literature reports²⁴⁻²⁶ for furan five:five and five:six ring system, respectively. Similarly 2e was reacted with vinyl ethyl ether and 3,4-dihydropyran under similar conditions to give good yields of the expected cyclic pyrazolines 4c and 5c, respectively.

⁽²¹⁾ Katritzky, A. R.; Rachwal, B.; Rachwal, S. J. Org. Chem. 1995, 60 3993

⁽²²⁾ Katritzky, A. R.; Nichols, D. A.; Qi, M.; Yang, B. J. Heterocycl. Chem. 1997, submitted.

⁽²³⁾ Katritzky, A. R.; Abonia, R.; Yang, B.; Qi, M.; Insuasty, B. Unpublished work

⁽²⁴⁾ Elslager, E. F.; Worth, D. F. J. Heterocycl. Chem. 1969, 6, 597. (25) Kametani, T.; Takeda, H.; Suzuki, Y.; Kasai, H.; Honda, T. Heterocycles 1986, 24, 3385.

⁽²⁶⁾ Kametani, T.; Furuyama, H.; Fukuoka, Y.; Takeda, H.; Suzuki, Y.; Honda, T. J. Heterocycl. Chem. 1986, 23, 185.

Scheme 3

Bt = Benzotriazolyl

Compound 5c also has the \emph{cis} configuration, which was confirmed by NOE, and the coupling constant of H-3 and H-4 is 3.8 Hz.

When excess of vinyl ethyl ether was reacted with 2a, 4-ethoxy- $\{1$ -phenyl-[2,3-a]-pyrazolido $\}$ tetrahydroquinoline ($\mathbf{8}$) was obtained in 31% yield. The formation of $\mathbf{8}$ can be explained as depicted in Scheme 3: N-(1-benzotriazolylmethyl)-N,N-diphenylhydrazine and vinyl ethyl ether in the presence of zinc bromide first gave compound $\mathbf{7}$, which is unstable under acidic conditions, and then reacted further with a second equivalent of vinyl ethyl ether followed by ring-closure to give compound $\mathbf{8}$. From the 1 H NMR of $\mathbf{8}$, the proton of the ring $\mathbf{4}$ -position CH to which the ethoxy group is attached is assigned to the signal at 4.45 ppm. This signal is a double-doublet with J=4.1 and 10.2 Hz. This means that there is one axial—axial coupling and thus that the ethoxy group is equatorial and trans to the proton at the ring junction.

Treatment of N-(1-benzotriazolylcyclohexylmethyl)-N,N-diphenylhydrazine (**2d**) with 3,4-dihydrofuran in the presence of zinc bromide gave 2-cyclohexyl-2,3,3a,4,5,9b-hexahydrofuro[3,2-c]quinoline (**9**) (39%), with cleavage of the N-N bond and the phenylamino group acting as a leaving group (Scheme 4). In the 1 H NMR spectrum of **9**, the doublet (J = 8.0 Hz) at 5.11 ppm was assigned to the 4-position proton. The coupling indicates²⁶ that the configuration between the 3- and 4-position protons is cis. Furthermore, the 2-position proton was at 3.11 ppm as a doublet (J = 2.5 Hz), which showed that the configuration between 2- and 3-position is also cis.²⁶

Treatment of **2f** with electron-rich alkene under the above conditions gave two products, **5d** and **10**, in 39% and 44%, respectively (Scheme 5). **5d** is also of *cis* configuration, as the coupling constant of H-3 and H-4 is 3.9 Hz. Compound **10** has a *trans* configuration, coupling constant of H-3 and H-4 being 10.2 Hz. In this reaction, the nucleophilicities of the amide nitrogen and *ortho*-carbon at the phenyl ring are comparable.

Scheme 4

Scheme 5

Ph H H N-N
$$C_2H_5$$
 C_2H_5 C_2H_5

In conclusion, N-(1-benzotriazolylalkyl)-N,N-disubstituted hydrazines ${\bf 2}$ are useful intermediates which provide convenient and novel methods for preparation of trisubstituted hydrazines ${\bf 3}$ and substituted pyrazolidines ${\bf 4}$ and ${\bf 5}$.

Experimental Section

General Comments. Melting points were determined with a Koefler hot-stage apparatus without correction. The 1H NMR and ^{13}C NMR spectra were recorded on a 300 MHz spectrometer in CDCl $_3$ with tetramethylsilane or the solvent as the internal reference. Tetrahydrofuran (THF) was distilled from sodium/benzophenone prior to use. Grignard reactions were carried out under an atmosphere of dry nitrogen. All glassware was oven-dried. All moisture-sensitive reagents were transferred by means of predried syringes.

General Procedure for the Synthesis of *N*-(1-Benzotriazolylalkyl)-*N*,*N*-disubstituted Hydrazines 2. Method A (2a-d). A mixture of benzotriazole (23.8 g, 0.2 mol), aldehyde (0.2 mol), and 1,2-diphenylhydrazine (18.4 g, 0.1 mol)

in methylene chloride (400 mL) was stirred at rt in the presence of molecular sieves (3 Å, 10 g) for an appropriate time until no hydrazine was left. The mixture was washed with agueous NaOH (2 N, 2×100 mL) and water (2×100 mL) and the aqueous phase was extracted with ethyl acetate. The combined organic layers were dried over anhydrous sodium sulfate. After the removal of the solvent, the residue was recrystallized from ether to give the expected compounds.

N-(Benzotriazol-1-ylmethyl)-N,N'-diphenylhydra**zine (2a)**: 1 H NMR (CDCl₃) δ 6.13 (s, 1 H), 6.29 (br s, 2 H), 6.87-6.94 (m, 4 H), 7.19-7.30 (m, 6 H), 7.34 (t, 1 H, J=7.5Hz), 7.43 (t, 1 H, J = 7.5 Hz), 7.54 (d, 1 H, J = 8.1 Hz), 8.02 (d, 1 H, J = 8.1 Hz); ¹³C NMR (CDCl₃) δ 64.0, 109.8, 112.8, 114.5, 120.0, 120.7, 121.2, 124.1, 127.8, 129.4, 129.6, 133.8, 145.9, 146.2, 146.9. Anal. Calcd for $C_{19}H_{17}N_5$: C, 72.35; H, 5.44; N, 22.22. Found: C, 72.32; H, 5.58; N, 22.45.

N-(1-Benzotriazolylpropyl)-N,N-diphenylhydrazine (2b): ¹H NMR (CDCl₃) δ (NMR data of Bt-2 isomer are listed in square brackets) 0.94 (t, 3 H, J = 7.4 Hz) [0.90 (t, 3 H, J =7.4 Hz], 2.52-2.65 (m, 2 H) [2.34-2.50 (m, 2 H)], 6.33 (br s,1 H), 6.57 (t, 1 H, J = 7.1 Hz) [6.70 (t, 1 H, J = 7.0 Hz)], 6.80– 7.47 (m, 12 H), 7.49 (d, 1 H, J = 8.5 Hz), 8.02 (d, 1 H, J = 8.2Hz) [7.82-7.86 (m, 2 H)]; ¹³C NMR (CDCl₃) δ 10.6 [10.2], 26.0 [26.9], 78.1 [82.8], 109.9, 112.3 [111.7], 116.3 [115.4], [118.2], [119.2], 119.6, 120.0, 122.0 [121.4], 123.9, [126.3], 127.5, 129.1, 129.3, 133.3, [143.7], 145.5, 147.7, 147.8 [148.2]. Anal. Calcd for C₂₁H₂₁N₅: C, 73,43; H, 6.17; N, 20.40. Found: C, 73.31; H, 6.41; N, 20.25.

N-(1-Benzotriazolylbutyl)-N,N-diphenylhydrazine (2c): ¹H NMR (CDCl₃) δ (NMR data of Bt-1 isomer are listed in square brackets, the ratio of Bt-2 to Bt-1 isomers is 5) 0.89 (t, 3° H, J = 7.5 Hz) [0.87 (t, 3 H, J = 7.3 Hz)], 1.10 - 1.30 (m, 1 H), 1.32-1.50 (m, 1 H), 2.28-2.41 (m, 1 H), 2.43-2.59 (m, 1 H), 6.27 (br s, 1 H), 6.66 (t, 1 H, J = 7.2 Hz), 6.75–7.47 (m, 12 H), 7.48 (d, 1 H, J = 8.2 Hz), 8.01 (d, 1 H, J = 8.2 Hz) [7.80– 7.86 (m, 2 H)]; 13 C NMR (CDCl₃) δ 13.4, 19.2 [18.8], 34.5 [35.3], 76.4 [81.2], 109.9, 112.3 [111.8], 116.4 [115.5], [118.2], [119.2], 119.7, 120.0, 122.0 [121.5], 123.9 [126.3], 127.5, 129.1, 129.3 [129.4], 133.2, [143.7], 145.6, 147.8 [147.7], 148.3. Anal. Calcd for C₂₂H₂₃N₅: C, 73,91; H, 6.49; N, 19.60. Found: C, 73.90; H, 6.53; N, 19.89.

N-(1-Benzotriazolyl-1-cyclohexylmethyl)-N,N-diphe**nylhydrazine (2d)**: 1 H NMR (CDCl₃) δ (NMR data of Bt-2 isomer are listed in square brackets) 0.58-0.70 (m, 1 H), 0.77- $1.45\ (m,\ 6\ H),\ 1.48 - \hat{1}.82\ (m,\ 3\ H),\ 2.18 - 2.30\ (m,\ 1\ H)\ [2.30 -$ 2.48 (m, 1 H)], 2.90-3.10 (m, 1 H) [2.69-2.86 (m, 1 H)], 6.41 (d, 1 H, J = 10.0 Hz) [6.41 (d, 1 H, J = 7.5 Hz)], 6.45 (br s, 1 H), 6.70-7.40 (m, 13 H), [7.80-7.85 (m, 2 H)], 7.97 (d, 1 H, J = 8.7 Hz); 13 C NMR (CDCl₃) δ 25.4 [25.3], 26.1 [26.0], 29.2 [28.8], 30.8 [30.6], 39.2 [39.7], 82.9 [87.7], 109.7, 112.5 [112.3], 112.8, 117.8 [117.7], 118.3 [119.1], [119.4], 119.8, 122.8 [122.1], 123.8, [126.3], 127.5, 129.1 [129.0], 129.2 [129.3], 131.2, [143.5], 145.2, 146.9 [147.4], 147.8 [149.0]. Anal. Calcd for $C_{25}H_{27}N_5$: C, 75.52; H, 6.85; N, 17.63. Found: C, 75.64; H, 6.98; N, 17.63.

Method B (for 2e,f). A mixture of benzotriazole (3.57 g, 0.03 mol), paraformaldehyde (0.90 g, 0.03 mol), and the substituted hydrazine (0.02 mol) was heated to reflux in benzene (100 mL) for 24 h with a Dean-Stark trap for removal of the formed water. The reaction mixture was washed with aqueous NaOH (1 N, 2×50 mL) and water (2×50 mL) and extracted with ethyl acetate. The combined organic layer was dried over anhydrous Na₂SO₄. After removal of the solvent, the residue was recrystallized from ethyl acetate and a white solid was obtained.

1,4-Diphenyl-4-(1-benzotriazol-1-ylmethyl)semicarba**zide (2e)**: 1 H NMR (CDCl₃) δ 6.21 (s, 1 H), 6.28 (s, 1 H), 6.99– 7.07 (m, 2 H), 7.18–7.37 (m, 10 H), 7.47 (t, 1 H, J = 7.5 Hz), 7.58 (d, 1 H, J = 8.40 Hz), 7.81 (s, 1 H), 8.01 (d, 1 H, J = 8.2Hz); 13 C NMR (CDCl₃) δ 67.2, 109.3, 115.6, 119.7, 120.3, 123.3, 123.6, 124.5, 128.5, 128.9, 129.8, 133.1, 137.5, 145.9, 146.2, 155.2. Anal. Calcd for $C_{20}H_{18}N_6O$: C, 67.01; H, 5.07; N, 23.46. Found: C, 67.22; H, 5.20; N, 23.23.

N-Phenyl-N-(benzotriazol-1-ylmethyl)-N-propionylhy**drazine (2f)**: ¹H NMR (CDCl₃) (at 70 °C) δ 1.08 (t, 3 H, J =7.5 Hz), 2.21 (q, 2 H, J = 7.2 Hz), 6.38 (s, 2 H), 6.86 (t, 1 H, J= 6.9 Hz), 7.12 (d, 2 H, J = 8.1 Hz), 7.27 (t, 2 H, J = 7.5 Hz), 7.41 (t, 1 H, J = 7.5 Hz), 7.55 (t, 1 H, J = 7.5 Hz), 7.78 (d, 1 H, J = 8.4 Hz), 8.04 (d, 1 H, J = 8.1 Hz); ¹³C NMR (CDCl₃) (at 70 °C) δ 9.4, 26.4, 63.8, 111.0, 112.8, 119.1, 119.7, 124.1, 127.4, 129.1, 132.9, 145.3, 146.6, 172.4. Anal. Calcd for C₁₆H₁₇N₅O: C, 65.05; H, 5.80; N, 23.72. Found: C, 65.31; H, 5.94; N, 23.93.

General Procedure for Preparation of Trisubstituted **Hydrazines 3a–g.** To a solution of N-(1-benzotriazolylalkyl)-N,N-diphenylhydrazine **2a**-**d** (5 mmol) in THF (50 mL) was added Grignard reagent (10 mmol) dropwise at 0 °C under nitrogen. The mixture was stirred at 0 °C for 1 h and then warmed to rt for 2 h. The reaction mixture was washed with aqueous NaOH solution (2 N, 2 \times 25 mL) and water (2 \times 25 mL) and extracted with ethyl acetate (2 \times 25 mL). The combined organic layers were dried over anhydrous sodium sulfate. After the removal of the solvent under vacuum, the residue was purified by column chromatography on silica gel to give the pure product.

N-Propyl-N,N-diphenylhydrazine (3a): oil; lit.²⁷ mp 32-34 °C; ¹H NMR (CDCl₃) δ 0.93 (t, 3 H, J = 7.5 Hz), 1.63-1.71 (m, 2 H), 3.42 (t, 2 H, J = 7.0 Hz), 5.53 (s, 1 H), 6.73-6.80 (m, 4 H), 6.81-6.90 (m, 2 H), 7.15-7.22 (m, 4 H); ¹³C NMR (CDCl₃) δ 11.5, 19.4, 53.9, 112.2, 112.8, 118.4, 119.4, 129.2, 129.3, 147.6, 149.8. Anal. Calcd for $C_{15}H_{18}N_2$: C, 79.59; H, 8.02; N, 12.38. Found: C, 79.49; H, 8.04; N, 12.61.

N-3-Pentyl-N,N-diphenylhydrazine (3b): 1H NMR $(CDCl_3)$ δ 0.75–1.14 (m, 6 H), 1.48–1.60 (m, 2 H), 1.62–1.75 (m, 2 H), 3.66 (quintet, 1 H, J = 6.3 Hz), 5.33 (s, 1 H), 6.72-6.83 (m, 4 H), 6.84-6.90 (m, 2 H), 7.12-7.22 (m, 4 H); ¹³C NMR (CDCl₃) δ 12.1, 24.3, 65.5, 112.2, 114.1, 118.8, 119.1, 129.2, 148.7, 150.8. Anal. Calcd for C₁₇H₂₂N₂: C, 80.26; H, 8.72; N, 11.02. Found: C, 80.51; H, 8.75; N, 10.92.

N-3-Hexyl-N,N-diphenylhydrazine (3c): 1H NMR (CDCl₃) δ 0.78–1.15 (m, 6 H), 1.16–1.80 (m, 6 H), 3.70–3.80 (m, 1 H), 5.31 (s, 1 H), 6.72-6.82 (m, 4 H), 6.84-6.90 (m, 2 H), 7.15-7.25 (m, 4 H); 13 C NMR (CDCl₃) δ 12.1, 14.2, 20.6, 33.7, 33.8, 63.2, 112.2, 114.1, 118.8, 119.1, 129.2, 148.7, 150.8. Anal. Calcd for C₁₈H₂₄N₂: C, 80.54; H, 9.01; N, 10.44. Found: C, 80.55; H, 9.07; N, 10.46.

N-(1-Ethyl-1-cyclohexylmethyl)-N,N-diphenylhydra**zine (3d)**: ¹H NMR (CDCl₃) δ 0.70–0.93 (m, 2 H), 1.00–1.30 (m, 6 H), 1.35-1.90 (m, 8 H), 3.48-3.58 (m, 1 H), 5.41 (s, 1 H), 6.70-6.90 (m, 6 H), 7.12-7.24 (m, 4 H); ¹³C NMR (CDCl₃) $\delta \ 12.7, \ 22.2, \ 26.3, \ 26.5, \ 32.0, \ 40.9, \ 69.2, \ 112.4, \ 113.4, \ 118.2,$ 119.1, 129.1, 129.2, 148.3, 152.0. Anal. Calcd for C₂₁H₂₈N₂: C, 81.77; H, 9.15; N, 9.08. Found: C, 81.70; H, 9.26; N, 9.08.

N-(α-Propylbenzyl)-*N,N*-diphenylhydrazine (3e): ¹H NMR (CDCl₃) δ 0.93 (t, 3 H, J = 7.5 Hz), 1.38–1.62 (m, 2 H), 1.88– 2.01 (m, 1 H), 2.07-2.19 (m, 1 H), 5.00 (t, 1 H, J = 7.4 Hz),5.09 (br s, 1 H), 6.71-6.83 (m, 4 H), 6.92-6.96 (m, 2 H), 7.10-7.28 (m, 9 H); 13 C NMR (CDCl₃) δ 14.1, 20.4, 33.6, 66.0, 112.3, 115.2, 119.1, 119.6, 127.6, 128.3, 129.2, 138.7, 148.5, 150.3. Anal. Calcd for $C_{22}H_{24}N_2$: C, 83.49; H, 7.65; N, 8.86. Found: C. 83.76; H, 7.90; N, 8.98.

N-3-Butenyl-N,N-diphenylhydrazine (3f): ¹H NMR (CDCl₃) δ 2.43 (q, 2 H, J = 6.9 Hz), 3.59 (t, 2 H, J = 6.9 Hz), 5.06 (d, 1 H, J = 11.4 Hz), 5.11 (d, 1 H, J = 17.1 Hz), 5.63 (s, 1 H), 5.77-5.95 (m, 1 H), 6.72-6.94 (m, 6 H), 7.14-7.27 (m, 4 H); 13 C NMR (CDCl₃) δ 30.9, 51.3, 112.4, 112.9, 116.9, 118.7, 119.7, 129.4, 129.5, 136.1, 147.6, 149.6. Anal. Calcd for C₁₆H₁₈N₂: C, 80.62; H, 7.62; N, 11.76. Found: C, 80.82; H, 7.25; N, 11.89.

N-4-(1-Heptenyl)-N,N-diphenylhydrazine (3g): 1H NMR $(CDCl_3) \delta 0.82-1.00 \text{ (m, 3 H)}, 1.20-1.80 \text{ (m, 4 H)}, 2.22-2.60$ (m, 2 H), 3.90-4.03 (m, 1 H), 5.01 (d, 1 H, J = 9.9 Hz), 5.08(d, 1 H, J = 17.3 Hz), 5.43 (br s, 1 H), 5.76-6.02 (m, 1 H), 6.75-6.92 (m, 6 H), 7.14-7.27 (m, 4 H); 13 C NMR (CDCl₃) δ 14.1, 20.5, 34.1, 36.2, 61.2, 112.2, 114.0, 116.6, 118.9, 119.2, 129.2, 136.7, 148.6, 150.3. Anal. Calcd for C₁₉H₂₄N₂: C, 81.37; H, 8.63; N, 10.00. Found: C, 81.20; H, 8.83; N, 9.98.

General Procedure for Preparation of Trisubstituted **Hydrazines 3h,i.** To a solution of furan (10 mmol, 0.8 mL) and TMEDA (11 mmol, 1.7 mL) in THF (50 mL) was added *n*-butyllithium (11 mmol, 1.6 M, 6.9 mL) dropwise at −78 °C

⁽²⁷⁾ Berg-Nielsen, K.; Bernatek, E. Acta Chem. Scand. 1972, 26, 4130.

under nitrogen. After stirring for 2 h, the reaction mixture was warmed to reflux for 0.5 h. The solution was then cooled to rt for 1 h, and anhydrous zinc bromide (11 mmol, 2.48 g) in THF (10 mL) was added. The mixture was stirred at rt for 1 h, and a solution of N-(benzotriazol-1-ylmethyl)-N,N-diphenylhydrazine (5 mmol, 1.58 g) in THF (5 mL) was added. The formed mixture was stirred at rt for 16 h. The reaction mixture was washed with aqueous NaOH solution (2 N, 2 × 25 mL) and water (2 × 25 mL) and extracted with ethyl acetate (2 × 25 mL). The combined organic layers were dried over anhydrous sodium sulfate. After the removal of the solvent under vacuum, the residue was purified by column chromatography on silica gel to give the pure product $\bf 3h$ (0.82 g, yield 62%).

N-2-Furyl-*N*,*N*-diphenylhydrazine (3h): ¹H NMR (CDCl₃) δ 4.67 (s, 2 H), 5.79 (s, 1 H), 6.21 (d, 1 H, J = 3.0 Hz), 6.30 (d, 1 H, J = 2.4 Hz), 6.80–6.85 (m, 2 H), 7.07 (d, 2 H, J = 8.1 Hz), 7.25 (t, 4 H, J = 7.5 Hz), 7.36 (s, 1 H); ¹³C NMR (CDCl₃) δ 47.3, 108.7, 110.2, 112.8, 113.4, 119.2, 120.0, 129.1, 129.4, 142.4, 147.1, 149.4, 150.7. Anal. Calcd for C₁₇H₁₆N₂O: C, 77.24; H, 6.11; N, 10.60. Found: C, 77.62; H, 6.33; N, 10.73.

N-2-Thiophenyl-*N*,*N*-diphenylhydrazine (3i): 1 H NMR (CDCl₃) δ 4.87 (s, 2 H), 5.64 (s, 1 H), 6.82–6.92 (m, 4 H), 6.93–6.98 (m, 2 H), 7.08 (d, 2 H, J = 6.3 Hz), 7.18–7.31 (m, 5 H); 13 C NMR (CDCl₃) δ 49.3, 113.0, 113.8, 119.5, 120.1, 125.4, 126.8, 127.0, 129.3, 129.5, 138.1, 147.0, 149.2. Anal. Calcd for C₁₇H₁₆N₂S: C, 72.83; H, 5.76; N, 10.00. Found: C, 72.71; H, 5.89; N, 10.01.

General Procedure for the Preparation of Pyrazolidine Derivatives 4 and 5 and Compounds 8–10. A mixture of N-(1-benzotriazolylalkyl)-N,N-disubstituted hydrazine 2 (2 mmol), alkene (4 mmol), and anhydrous zinc bromide (20 mg) in dry methylene chloride (30 mL) was stirred at rt for an appropriate time. After the reaction was completed, the mixture was washed with aqueous NaOH (2 N, 2 \times 15 mL) and water (2 \times 15 mL) and extracted with ethylacetate. The combined organic layers were dried over anhydrous sodium sulfate. After the removal of the solvent under vacuum, the residue was purified by column chromatography on silica gel to give the pure products.

3-Pyrrolidonyl-1,2-diphenylpyrazolidine (4a): ¹H NMR (CDCl₃) δ 1.85–1.98 (m, 2 H), 2.17–2.26 (m, 1 H), 2.36–2.48 (m, 3 H), 3.06–3.16 (m, 1 H), 3.22–3.32 (m, 1 H), 3.56–3.68 (m, 1 H), 3.89–3.98 (m, 1 H), 6.11 (t, 1 H, J= 6.5 Hz), 6.84–6.96 (m, 4 H), 7.06 (d, 2 H, J= 8.1 Hz), 7.23–7.32 (m, 4 H); ¹³C NMR (CDCl₃) δ 18.1, 30.4, 31.6, 43.7, 49.1, 72.9, 113.8, 114.5, 119.7, 120.9, 129.1, 149.8, 150.0, 174.9. Anal. Calcd for C₁₉H₂₁N₃O: C, 74.23; H, 6.89; N, 13.68. Found: C, 74.50; H, 7.02; N, 13.77.

1,2-Diphenyl-3-(*N*-methylacetamidyl)pyrazolidine (4b): $^{1}H \ NMR \ (CDCl_{3}) \ \delta \ 2.01-2.08 \ (m, 1\ H), \ 2.15 \ (s, 3\ H), \ 2.39-2.46 \ (m, 1\ H), \ 2.76 \ (s, 3\ H), \ 3.46-3.54 \ (m, 1\ H), \ 3.96-4.08 \ (m, 1\ H), \ 6.54 \ (t, 1\ H, \ \emph{\textit{\textit{J}}}=4.5\ Hz), \ 6.84-6.98 \ (m, 6\ H), \ 7.22-7.28 \ (m, 4\ H); \ ^{13}C \ NMR \ (CDCl_{3}) \ \delta \ 22.7, \ 30.8, \ 31.0, \ 48.8, \ 75.2, \ 113.9, \ 114.1, \ 119.5, \ 120.6, \ 129.1, \ 149.9 \ (overlapped), \ 170.8. \ Anal.
Calcd for $C_{18}H_{21}N_{3}O$: \$C, 73.18; \$H, 7.17; \$N, 14.23. Found: \$C, 73.48; \$H, 7.35; \$N, 14.37.}

1-Phenyl-2-(phenylcarbamoyl)-3-ethoxypyrazolidine (4c): 1 H NMR (CDCl₃) δ 1.16 (t, 3 H, J = 6.9 Hz), 2.12–2.26 (m, 1 H), 2.29–2.40 (m, 1 H), 3.52–3.68 (m, 1 H), 3.70–3.85 (m, 2 H), 3.86–4.00 (m, 1 H), 5.81–5.87 (m, 1 H), 6.88–7.10 (m, 2 H), 7.15 (d, 2 H, J = 8.2 Hz), 7.22–7.36 (m, 4 H), 7.45 (d, 2 H, J = 8.0 Hz), 8.14 (br s, 1 H); 13 C NMR (CDCl₃) δ 14.8, 33.6, 53.2, 63.9, 88.4, 115.5, 119.2, 122.0, 123.3, 128.8, 129.0, 138.0, 150.9, 156.6. Anal. Calcd for C₁₈H₂₁N₃O₂: C, 69.42; H, 6.80; N, 13.50. Found: C, 69.40; H, 7.15; N, 13.53.

N,N-Diphenyl-2,3,3a,4,5,6a-hexahydrofuro[2,3-c]pyrazole (5a): 1 H NMR (CDCl₃) δ 1.58–1.69 (m, 1 H), 1.97–2.22 (m, 1 H), 3.07–3.17 (m, 1 H), 3.69–3.91 (m, 4 H), 5.83 (d, 1 H, J = 6.1 Hz), 6.80–6.92 (m, 2 H), 6.92–7.00 (m, 2 H), 7.00–7.10 (m, 2 H), 7.18–7.30 (m, 4 H); 13 C NMR (CDCl₃) δ 31.2, 46.0, 56.7, 67.5, 100.0, 113.4, 114.0, 119.5, 120.1, 128.9, 129.1, 147.8, 150.6. Anal. Calcd for C₁₇H₁₈N₂O: C, 76.65; H, 6.82; N, 10.52. Found: C, 76.93; H, 7.13; N, 10.57.

N,N-Diphenyl-1,2,3,3a,4,5,6,7a-octahydropyrano[2,3-c]pyrazole (5b): ¹H NMR (CDCl₃) δ 1.38–1.48 (m, 1 H), 1.65–1.88 (m, 3 H), 2.46–2.58 (m, 1 H), 3.48–3.61 (m, 2 H), 3.82–3.99 (m, 2 H), 5.12 (d, 1 H, J = 2.3 Hz), 6.80–6.98 (m, 4 H), 7.10–7.32 (m, 6 H); ¹³C NMR (CDCl₃) δ 20.9, 22.0, 36.8, 52.5, 64.8, 94.3, 113.6, 115.9, 119.0, 121.3, 128.9, 149.2, 151.5. Anal. Calcd for C₁₈H₂₀N₂O: C, 77.10; H, 7.19; N, 10.00. Found: C, 77.27; H, 7.50; N, 10.02.

2-Phenyl-1-(phenylcarbamoyl)-1,2,3,3a,4,5,6,7a-octahydropyrano[2,3-*c***]pyrazole (5c)**: 1 H NMR (CDCl₃) δ 1.29–1.40 (m, 1 H), 1.53–1.68 (m, 1 H), 1.77–1.96 (m, 2 H), 2.44–2.58 (m, 1 H), 3.45 (q, 2 H, J = 9.90 Hz), 3.80–4.00 (m, 2 H), 5.74 (d, 1 H, J = 3.8 Hz), 6.95–7.12 (m, 4 H), 7.20–7.33 (m, 4 H), 7.34–7.43 (m, 2 H), 7.97 (br s, 1 H); 13 C NMR (CDCl₃) δ 19.9, 21.5, 38.3, 55.3, 65.6, 85.8, 115.0, 119.3, 121.6, 123.3, 128.7, 129.0, 137.9, 152.1, 155.6. Anal. Calcd for C₁₉H₂₁N₃O₂: C, 70.55; H, 6.55; N, 13.00. Found: C, 70.75; H, 6.74; N, 13.04.

2-Phenyl-1-propionyl-1,2,3,3a,4,5,6,7a-octahydropyra-no[2,3-*c***]pyrazole (5d)**: 1 H NMR (CDCl₃) δ 0.97 (t, 3 H, J = 7.5 Hz), 1.26–1.38 (m, 1 H), 1.47–1.58 (m, 1 H), 1.62–1.72 (m, 1 H), 2.14–2.28 (m, 1 H), 2.32–2.50 (m, 1 H), 3.32–3.50 (m, 2 H), 3.66–3.84 (m, 2 H) H), 5.58 (d, 1 H, J = 3.9 Hz), 6.82–6.90 (m, 3 H), 7.20 (t, 2 H, J = 8.1 Hz); 13 C NMR (CDCl₃) δ 7.9, 19.2, 20.4, 25.2, 36.8, 55.2, 63.7, 83.2, 114.9, 120.0, 127.8, 151.8, 173.6. Anal. Calcd for C₁₅H₂₀N₂O₂: C, 69.19; H, 7.75; N, 10.77. Found: C, 69.21; H, 8.02; N, 10.75.

4-Ethoxy-(1,2,3,3a,4,5-hexahydro-1-phenylpyrazolo[1,5-a]quinoline (8): mp 113–115 °C; ¹H NMR (CDCl₃) δ 1.35 (t, 3 H, J = 6.9 Hz), 1.54–1.69 (m, 1 H), 1.88–1.98 (m, 1 H), 2.28–2.38 (m, 1 H), 2.52–2.60 (m, 1 H), 3.49–3.88 (m, 5 H), 4.45 (dd, 1 H, J = 4.1, 10.2 Hz), 6.82–6.97 (m, 5 H), 7.10 (t, 1 H, J = 7.5 Hz), 7.24 (t, 2 H, J = 7.8 Hz), 7.37 (d, 1 H, J = 7.5 Hz); 13 C NMR (CDCl₃) δ 15.6, 32.6, 34.0, 51.1, 55.0, 65.0, 72.9, 113.7, 114.0, 119.0, 119.7, 123.8, 126.2, 127.9, 128.9, 144.7, 151.5. Anal. Calcd for $C_{19}H_{22}N_2O$: C, 77.51; H, 7.54; N, 9.52. Found: C, 77.36; H, 7.72; N, 9.53.

2-Cyclohexyl-2,3,3a,4,5,9b-hexahydrofuro[3,2-c]quinoline (9): mp 148–151 °C; ¹H NMR (CDCl₃) δ 0,94–1.08 (m, 2 H), 1.15–1.49 (m, 4 H), 1.60–2.10 (m, 7 H), 2.74 (q, 1 H, J= 8.1 Hz), 3.11 (dd, 1 H, J= 2.5, 9.1 Hz), 3.70–3.82 (m, 3 H), 5.11 (d, 1 H, J= 8.0 Hz), 6.52 (d, 1 H, J= 7.8 Hz), 6.74 (t, 1 H, J= 7.5 Hz), 7.04 (t, 1 H, J= 8.2 Hz), 7.29 (d, 1 H, J= 7.5 Hz); 13 C NMR (CDCl₃) δ 23.9, 25.9, 26.0, 26.3, 28.9, 30.1, 40.3, 40.4, 57.7, 66.5, 76.0, 114.4, 118.6, 122.8, 128.1, 130.0, 145.1. Anal. Calcd for C₁₇H₂₃N₁O: C, 79.32; H, 9.01; N, 5.44. Found: C, 79.13; H, 9.14; N, 5.44.

N-(Propionylamino)-3,4,4a,5,6,10b-hexahydropyrano[3,2-c]quinoline (10): yield 44%, mp 166–168 °C; ¹H NMR (CDCl₃) δ 1.12 (t, 3 H, J= 7.2 Hz), 1.42–1.54 (m, 1 H), 1.62–1.84 (m, 2 H), 1.84–1.98 (m, 2 H), 2.17–2.29(m, 2 H), 2.49–2.54 (m, 1 H), 3.19 (d, 1 H, J= 10.2 Hz), 3.58–3.72 (m, 2 H), 3.81 (d, 1 H, J= 10.2 Hz), 4.44 (s, 1 H), 6.65–6.75 (m, 2 H), 7.07–7.19 (m, 2 H), 9.09 (s, 1 H); ¹³C NMR (CDCl₃) δ 8.8, 21.8, 24.4, 26.2, 31.7, 50.0, 65.7, 72.5, 111.3, 117.1, 120.9, 128.0, 129.6, 145.3, 171.4. Anal. Calcd for C₁₇H₂₃N₁O: C, 69.19; H, 7.75; N, 10.77. Found: C, 69.23; H, 8.12; N, 10.22.

JO9709110